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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.005 Å R factor = 0.054 wR factor = 0.155 Data-to-parameter ratio = 15.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# (*R*,*R*)-*N*,*N*'-Bis(5-chlorosalicylidene)-1,2-cyclohexanediamine

In the title compound {systematic name: (R,R)-4,4'-dichloro-2,2'-[cyclohexane-1,2-diylbis(nitrilomethylidyne)]diphenol},  $C_{20}H_{20}Cl_2N_2O_2$ , there are two chiral C atoms and two intramolecular O-H···N hydrogen bonds. The molecule lies on a twofold rotation axis. The crystal structure is stabilized by intermolecular O-H···O hydrogen bonds, which link the molecules into one-dimensional helical chains along the *b* axis.

#### Comment

Chiral salen compounds are important chiral ligands widely used in asymmetric catalytic synthesis (Canail & Sherrington, 1999; Jacobsen, 2000). The structure of chiral salen compounds has a crucial effect on enantioselectivity and activity in asymmetric catalytic reactions (Nicewicz *et al.*, 2004; Yao *et al.*, 2001). Our research is focused on asymmetric synthesis catalysed by chiral salen-metal complexes (Zhu *et al.*, 2004). In order to study the relationship between the structures and properties of such salen compounds, we have synthesized the chiral ligand (R,R)-N,N'-bis(5-chloro-salicylidene)-1,2-cyclohexanediamine, (I), and present its crystal structure here.



The molecular structure of (I) (Fig. 1) contains two chiral C atoms in (R,R)-diastereomeric form, the molecule lying on a twofold rotation axis. Intramolecular O-H···N hydrogen bonds (Fig. 1, Table 1) are present.

The crystal packing is stabilized by intermolecular O– $H\cdots$ O hydrogen bonds (Table 1), which link the molecules into one-dimensional helical chains along the *b* axis (Fig. 2).

### **Experimental**

Under nitrogen, a mixture of (R,R)-1,2-cyclohexanediamine (342 mg, 3 mmol), Na<sub>2</sub>SO<sub>4</sub> (2 g) and 5-chloro-2-hydroxybenzaldehyde (939 mg, 6 mmol) in absolute ethanol (10 ml) was refluxed for about 12 h to yield a yellow precipitate. The product was collected by vacuum filtration and washed with ethanol. The crude solid was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (50 ml) and washed with water (2 × 10 ml) and brine (10 ml). After drying over Na<sub>2</sub>SO<sub>4</sub>, the solvent was removed under vacuum and a yellow solid was isolated in 85% yield (1.0 g). Yellow single crystals of (I) suitable for X-ray analysis were grown from a solution in hexane by slow evaporation of the solvent at room temperature over a period of about a week. Spectroscopic analysis:

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#### Figure 1

The molecular structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii. [Symmetry code: (i) 1 - x, 1 - y, z.]

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>,  $\delta$ , p.p.m.): 13.13 (*s*, 2H), 8.10 (*s*, 2H), 7.10–7.14 (*dd*, *J* = 2.0 and 8.8 Hz, 2H), 6.98–7.04 (*d*, *J* = 2.0 Hz, 2H), 6.75–6.78 (*d*, *J* = 8.8 Hz, 2H), 3.23–3.26 (*m*, 2H), 1.82–1.87 (*m*, 4H), 1.62–1.65 (*m*, 2H), 1.36–1.42 (*m*, 2H); IR (KBr,  $\nu$ , cm<sup>-1</sup>): 3430, 2924, 2856, 1633, 1478, 1371, 1282, 1185, 1293, 1034, 977; analysis calculated for C<sub>20</sub>H<sub>20</sub>Cl<sub>2</sub>N<sub>2</sub>O<sub>2</sub> (%): C 61.39, H 5.15, N 7.16; found: C 61.28, H 5.24, N 7.22.

#### Crystal data

 $C_{20}H_{20}Cl_2N_2O_2$  1

  $M_r = 391.28$  0

 Orthorhombic,  $P2_12_12$  a = 18.990 (6) Å

 b = 5.829 (2) Å
 b = 5.829 (2) Å

 c = 8.839 (3) Å
 d = 2 

 V = 978.4 (6) Å<sup>3</sup>
 d = 2 

  $D_x = 1.328$  Mg m<sup>-3</sup>
 d = 1328 Mg m<sup>-3</sup>

 Data collection
 Bruker SMART APEX CCD areadetector diffractometer

 $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Bruker, 2000)  $T_{\min} = 0.91, T_{\max} = 0.93$ 5037 measured reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.054$   $wR(F^2) = 0.155$  S = 1.051879 reflections 118 parameters H-atom parameters constrained Mo  $K\alpha$  radiation Cell parameters from 780 reflections  $\theta = 2.5-25.0^{\circ}$  $\mu = 0.35 \text{ mm}^{-1}$ T = 293 (2) K Block, yellow  $0.30 \times 0.24 \times 0.22 \text{ mm}$ 

1879 independent reflections 1554 reflections with  $l > 2\sigma(l)$   $R_{int} = 0.058$   $\theta_{max} = 26.0^{\circ}$   $h = -21 \rightarrow 23$   $k = -5 \rightarrow 7$  $l = -10 \rightarrow 10$ 

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0882P)^{2} + 0.127P]$ where  $P = (F_{o}^{2} + 2F_{c}^{2})/3$   $(\Delta/\sigma)_{max} < 0.001$   $\Delta\rho_{max} = 0.34 \text{ e} \text{ Å}^{-3}$   $\Delta\rho_{min} = -0.20 \text{ e} \text{ Å}^{-3}$ Absolute structure: Flack (1983),
with 732 Friedel pairs
Flack parameter = 0.13 (15)





The one-dimensional hydrogen-bonded helical chain in (I), viewed down the c axis.

# Table 1Hydrogen-bonding geometry (Å, °).

| $D - H \cdot \cdot \cdot A$         | $D-\mathrm{H}$ | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - \mathbf{H} \cdots A$ |
|-------------------------------------|----------------|-------------------------|--------------|---------------------------|
| $O1 - H1B \cdot \cdot \cdot N1$     | 0.85           | 2.03                    | 2.599 (4)    | 124                       |
| $O1 - H1B \cdot \cdot \cdot O1^{i}$ | 0.85           | 2.46                    | 2.903 (5)    | 114                       |

Symmetry code: (i) 1 - x, 1 - y, z.

All H atoms were positioned geometrically and refined using a riding model, with C–H distances in the range 0.93–0.98 Å and an O–H distance of 0.85 Å, and with  $U_{iso}(H) = 1.2U_{eq}(C,O)$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve structure: *SHELXTL* (Bruker, 2000); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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