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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.054$
$w R$ factor $=0.155$
Data-to-parameter ratio $=15.9$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## ( $R, R$ )- $N, N^{\prime}$-Bis(5-chlorosalicylidene)-1,2-cyclohexanediamine

In the title compound \{systematic name: $(R, R)$-4,4'-dichloro-2,2'-[cyclohexane-1,2-diylbis(nitrilomethylidyne)]diphenol\}, $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}$, there are two chiral C atoms and two intramolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds. The molecule lies on a twofold rotation axis. The crystal structure is stabilized by intermolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds, which link the molecules into one-dimensional helical chains along the $b$ axis.

## Comment

Chiral salen compounds are important chiral ligands widely used in asymmetric catalytic synthesis (Canail \& Sherrington, 1999; Jacobsen, 2000). The structure of chiral salen compounds has a crucial effect on enantioselectivity and activity in asymmetric catalytic reactions (Nicewicz et al., 2004; Yao et al., 2001). Our research is focused on asymmetric synthesis catalysed by chiral salen-metal complexes (Zhu et al., 2004). In order to study the relationship between the structures and properties of such salen compounds, we have synthesized the chiral ligand $(R, R)-N, N^{\prime}$-bis(5-chloro-salicylidene)-1,2-cyclohexanediamine, (I), and present its crystal structure here.

(I)

The molecular structure of (I) (Fig. 1) contains two chiral C atoms in $(R, R)$-diastereomeric form, the molecule lying on a twofold rotation axis. Intramolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds (Fig. 1, Table 1) are present.

The crystal packing is stabilized by intermolecular $\mathrm{O}-$ $\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Table 1), which link the molecules into one-dimensional helical chains along the $b$ axis (Fig. 2).

## Experimental

Under nitrogen, a mixture of ( $R, R$ )-1,2-cyclohexanediamine ( 342 mg , $3 \mathrm{mmol}), \quad \mathrm{Na}_{2} \mathrm{SO}_{4}(2 \mathrm{~g})$ and 5-chloro-2-hydroxybenzaldehyde ( $939 \mathrm{mg}, 6 \mathrm{mmol}$ ) in absolute ethanol ( 10 ml ) was refluxed for about 12 h to yield a yellow precipitate. The product was collected by vacuum filtration and washed with ethanol. The crude solid was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{ml})$ and washed with water $(2 \times 10 \mathrm{ml})$ and brine ( 10 ml ). After drying over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, the solvent was removed under vacuum and a yellow solid was isolated in $85 \%$ yield $(1.0 \mathrm{~g})$. Yellow single crystals of (I) suitable for X-ray analysis were grown from a solution in hexane by slow evaporation of the solvent at room temperature over a period of about a week. Spectroscopic analysis:


Figure 1
The molecular structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level and H atoms are shown as small spheres of arbitrary radii. [Symmetry code: (i) $1-x, 1-y, z$.]
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$, $\delta$, p.p.m.): $13.13(s, 2 \mathrm{H}), 8.10(s, 2 \mathrm{H})$, $7.10-7.14(d d, J=2.0$ and $8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.98-7.04(d, J=2.0 \mathrm{~Hz}, 2 \mathrm{H})$, $6.75-6.78(d, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.23-3.26(m, 2 \mathrm{H}), 1.82-1.87(m, 4 \mathrm{H})$, 1.62-1.65 ( $m, 2 \mathrm{H}$ ), 1.36-1.42 ( $m, 2 \mathrm{H}$ ); IR (KBr, $v, \mathrm{~cm}^{-1}$ ): 3430, 2924, 2856, 1633, 1478, 1371, 1282, 1185, 1293, 1034, 977; analysis calculated for $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}$ (\%): C 61.39, H 5.15, N 7.16; found: C 61.28, H 5.24, N 7.22 .

## Crystal data

$\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}$
$M_{r}=391.28$
Orthorhombic, $P 2_{1} 2_{1} 2$
$a=18.990$ (6) $\AA$
$b=5.829$ (2) $\AA$
$c=8.839$ (3) $\AA$
$V=978.4(6) \AA^{3}$
$Z=2$
$D_{x}=1.328 \mathrm{Mg} \mathrm{m}^{-3}$

## Data collection

Bruker SMART APEX CCD areadetector diffractometer $\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Bruker, 2000)
$T_{\text {min }}=0.91, T_{\text {max }}=0.93$
5037 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.054$
$w R\left(F^{2}\right)=0.155$
$S=1.05$
1879 reflections
118 parameters
H -atom parameters constrained

Mo $K \alpha$ radiation
Cell parameters from 780
reflections
$\theta=2.5-25.0^{\circ}$
$\mu=0.35 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, yellow
$0.30 \times 0.24 \times 0.22 \mathrm{~mm}$

1879 independent reflections
1554 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.058$
$\theta_{\text {max }}=26.0^{\circ}$
$h=-21 \rightarrow 23$
$k=-5 \rightarrow 7$
$l=-10 \rightarrow 10$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0882 P)^{2}\right. \\
& +0.127 P \text { ] } \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\text {max }}=0.34 \mathrm{e}^{\mathrm{\circ}}{ }^{-3} \\
& \Delta \rho_{\min }=-0.20 \mathrm{e}^{-3} \\
& \text { Absolute structure: Flack (1983), } \\
& \text { with } 732 \text { Friedel pairs } \\
& \text { Flack parameter }=0.13(15)
\end{aligned}
$$



Figure 2
The one-dimensional hydrogen-bonded helical chain in (I), viewed down the $c$ axis.

Table 1
Hydrogen-bonding geometry $\left(\AA^{\circ},^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| O1-H1B $\cdots \mathrm{N} 1$ | 0.85 | 2.03 | $2.599(4)$ | 124 |
| ${\text { O1-H1 } B \cdots 1^{\mathrm{i}}}^{1}$ | 0.85 | 2.46 | $2.903(5)$ | 114 |

Symmetry code: (i) $1-x, 1-y, z$.
All H atoms were positioned geometrically and refined using a riding model, with $\mathrm{C}-\mathrm{H}$ distances in the range $0.93-0.98 \AA$ and an $\mathrm{O}-\mathrm{H}$ distance of $0.85 \AA$, and with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C}, \mathrm{O})$.

Data collection: SMART (Bruker, 2000); cell refinement: SMART; data reduction: SAINT (Bruker, 2000); program(s) used to solve structure: SHELXTL (Bruker, 2000); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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